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## Preliminary Phytochemical Analysis of *Solanum nigrum* L. Variant Black and Orange Fruited

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### Abstract

The medicinal plant *Solanum nigrum* L., popularly known as black nightshade belongs to the family Solanaceae was selected for preliminary phytochemical screening of secondary metabolites. Phytochemical constituents like alkaloids, flavonoids, carbohydrates and phytosterols were analyzed. The alkaloids in black fruited and orange fruited variants of *Solanum nigrum* was found to  $48.61 \pm 0.83$  mg/g  $35.65 \pm 0.95$  mg/g respectively. The present study suggest that *Solanum nigrum* variants might be a source of large amount of alkaloids. The results tend to confirm the popular use of the plant.

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### Introduction

The medicinal plants occupy a significant place in modern medicine as raw materials for some important drugs, although synthetic drugs and antibiotics brought about an evolution in controlling different diseases. But these synthetic drugs are out of reach for millions of people. Those who live in remote places depend on traditional healers, whom they know and trust. The judicious use of medicinal herbs can even cure deadly diseases that have long defined synthetic drugs (Bhattacharjee, 2001). There are many plant herbs having different medicinal value used against various diseases since early time. Among them *Solanum nigrum* L. (Solanaceae) popularly known as black nightshade is

one of the plants having great importance in Ayurvedic medication. The member of this family known for the presence of natural products of medicinal significance mainly steroidal lactones, glycosides, alkaloids and flavonoids. The plant has been extensively used in traditional medicine in India and other parts of world to cure liver disorders, chronic skin ailments, inflammatory conditions, fevers, diarrhoea, ulcer etc., (Kritikar and Basu, 1935).

In the literature no work has been carried out in *Solanum nigrum* L. variants of black and orange fruited plants comparatively. Therefore the present study was aimed at evaluate preliminary phytochemical screening of variants of *Solanum nigrum*.

## Materials and methods

### Collection of plant sample

The whole plants of *Solanum nigrum* L. variants black fruited and orange fruited were collected from Thirumalairayan Pattinam 609 606, Karaikal Region, Pondicherry, Union Territory of South India, during March – April 2011. Both plant samples were collected during same season and same time (evening).

### Plant identification

The identity of the plant specimens were confirmed by using the Floras (Gamble, 1957; Matthew, 1983; Nair and Henry, 1983) and other treatises (Anonymous, 1992; Chatterjee and Pakrashi, 1994; Kirtikar and Basu, 1935). The botanical identity was also authenticated by Dr.M.Jegadeesan, Professor and Head, Department of Environmental and Herbal Sciences, Tamil University, Thanjavur, Tamil Nadu, India. Identity of the plant was confirmed with the help of type specimens available in the Herbarium of Botanical Survey of India, Southern Circle, TNAU Campus, Coimbatore, Tamil Nadu. The Herbarium specimens were prepared following the method of Jain and Rao (1976). The herbarium number in BSI is “BSI/SRC/5/23/2012-13/Tech.1480”. The herbarium specimen was deposited at Tamil University Herbarium (TUH-300).

### Preparation of powder (Harborne, 1973)

The shade dried entire plant powders of two plants were mechanically ground to coarse powder and passed through a 80 mesh sieve and used for further microscopical, physicochemical, phytochemical and fluorescence analysis.

### Fluorescence analysis (Kokoshi et al., 1958)

Fluorescence analysis of the plants powders were observed under day light and ultraviolet (UV) light (254 nm). Behaviours of powdered plant materials with different chemical reagents are identified.

### Physicochemical studies (Anonymous, 1996)

#### Determination of moisture content (Loss on drying)

Two grams of the powdered samples were taken in a

tared weighing bottle and weighed accurately. Dried at 105°C for 5 hrs and allowed to cool in a desiccator and weighed. The drying was continued at 150°C and weighed at 1 hr intervals. When the weight of the sample became constant, the loss in weight and the percentage of loss on drying were calculated.

#### Determination of total ash value

Two grams of powdered samples were taken in a tared silica dishes previously ignited and weighed; the ground powders were scattered in a fine even layer at the bottom of the dish and gradually by increasing heat (electrical furnace) not exceeding dull red by heating until free from carbon, cooled and weighed. As a carbon free form was not obtained, the charred mass was digested with hot water and the residue was collected in an ashless filter paper. The residue with the filter paper was incinerated and filtrate was added, evaporated to dryness and ignited at a low temperature. The percentage of ash with reference to the air dried powder was calculated.

#### Acid insoluble ash value

Total ash obtained was heated with addition of 25 ml of diluted HCl for 10 minutes. It was filtered in an ashless filter paper (Whatman No.41) and the residue was ignited in the furnace to get a constant weight.

#### Sulphated ash value

The ash of the powders was moistened with 1ml of H<sub>2</sub>SO<sub>4</sub> and ignited to 800 ± 25°C until reaching a constant weight.

#### Phytochemical studies

##### Solubility percentage (Anonymous, 1985)

##### In alcohol

Five grams of the air-dried powders were macerated with 100 ml of alcohol of the specified strength and kept in a closed flask for 24 hrs and frequently shaken for 6 hrs and allowed to stand for 18 hrs. Then filtered rapidly with precautions against the loss of alcohol, 25 ml of the filtrate was evaporated to dryness in a tared flat-bottomed shallow dish, dried at 105°C and weighed. The percentage of alcohol-soluble extractive with reference to the air dried samples was calculated.

## **In water**

The above said procedure was followed but water was used for extraction instead of alcohol. The extract was weighed and its percentage was calculated in terms of the air-dried weight of the plant material. The colour and consistency of the extracts were also noted.

## **Preparation of extracts (Anonymous, 1996)**

The plant powders of two variants of *Solanum nigrum* were extracted successively using Soxhlet apparatus with petroleum ether (60-80°C), hexane (63-70°C) chloroform (60°C) and alcohol (78°C). Each time before extracting with next solvent, powdered material were dried in an air-oven below 50°C. The extracts were dried over anhydrous sodium sulfate, stored in sealed vials in refrigerator (5-8°C) until analysis.

Finally, marc was macerated with chloroform water for 24 hrs to obtain the aqueous extract. The extract was concentrated by distilling off the solvent and then evaporating to dryness on a water bath.

## **Organoleptic characters**

Characters such as colour, odour, taste and texture of the two plant extracts were observed.

## **Qualitative phytochemical studies**

Qualitative phytochemical analysis was carried out according to Kokate et al. (1995). Carbohydrates, alkaloids, fixed oils and fats, saponins, tannins, phenols and terpenes were qualitatively analysed.

### **Alkaloids**

The extracts were dissolved in dilute H<sub>2</sub>SO<sub>4</sub> and filtrate was treated with Dragendorff's, Hager's, Mayer's and Wagner's reagent separately. Appearance of orange brown, yellow cream, pink and reddish brown precipitates in response to the above reagents respectively indicates the presence of alkaloids.

### **Carbohydrates**

The extracts were treated with Benedict's and Fehling's reagents under suitable conditions. Appearance of brick red colour in response to the above reagents indicates the presence of carbohydrates.

## **Fixed oils and fats**

A drop of concentrated extract was pressed in between two filter papers and kept undisturbed. The oil stains on the paper indicates the presence of oils and fats.

## **Saponins**

About 1 ml of the alcoholic and aqueous extracts was dissolved separately in 20 ml of water and shaken in a graduated cylinder for 15 minutes. Formation of 1 cm layer of foam indicates the presence of saponins.

## **Tannins and phenols**

Small quantity of alcoholic and aqueous extracts were dissolved in water and to that FeCl<sub>3</sub> (5%) or gelatine solution (1%) or lead acetate solution (10%) was added. Appearance of blue colour with FeCl<sub>3</sub> (or) precipitation with other reagents indicates the presence of tannins and phenols.

## **Gums and mucilage**

Add about 10 ml of aqueous extract slowly to 25 ml of absolute alcohol with constant stirring. Precipitation indicates the presence of gums and mucilages.

## **Quantitative phytochemical studies**

### **Estimation of total alkaloids (Ferguson, 1956)**

The plant powder weighing 100 g was taken separately and soaked in alcohol for 24 hrs. Then filtered, the filtrant was extracted with 0.1 N HCl and partitioned with chloroform in a separating funnel.

The chloroform layer was rejected and the aqueous layer basified with ammonium hydroxide to pH and partitioned with chloroform in a separating funnel. The aqueous layer was rejected and the chloroform layer was evaporated, the resultant content was treated as total alkaloid.

### **Estimation of total terpenoids**

One hundred gram of plant powder was taken separately and soaked in alcohol for 24 hrs. Then filtered, the filtrant was extracted with petroleum ether and the resultant content was treated as total terpenoid.

### Estimation of tannin-free total glycoside (Jainu et al., 2006)

Air dried 100 g powder was extracted with ethanol water (2:1). The aqueous ethanol extracts thus obtained contains tannins which usually interfere with biological activities. Hence, this should be removed by treating with 5% neutral lead acetate reagent which precipitates the tannins as lead tannate. The aqueous ethanolic solution was treated with 8% neutral lead acetate solution and the precipitated lead tannate was filtered off. This process is repeated with until no more precipitated was obtained. The clear filtrate now contained the excess unprecipitated lead ions in solutions which were removed by passing H<sub>2</sub>S gas into the solution. This removes the lead ions as insoluble complex black lead sulphide. The black precipitate was filtered and this process was usually reported until no more black precipitate was formed and the solution strongly smelled of H<sub>2</sub>S. The solution, usually of syrupy consistency was concentrated over water bath maintained at 55°C. This procedure was to remove the excess of H<sub>2</sub>S.

### Estimation of total flavonoid (Jainu et al., 2006)

Isolation of flavonoid from ethanol extract of any plant sample (powder) was carried out on the basis of solubility. For isolation, distilled water (100ml) was added to the concentrate of ethanol extract (50ml). After 1 hr precipitation was observed. This precipitate was recovered by filtration. Further, the precipitate was dissolved in chloroform (100ml) by shaking for 15 minutes and heated gently for 5 minutes and filtered in hot state. The chloroform soluble fraction was discarded and insoluble fraction, left on filter paper was dissolved in ethyl acetate, crystallized with methanol, thereafter

the residue obtained. For characterization positive result for Shinoda test which is characteristic of flavonoids.

### Results

Table 1 shows the results of organoleptic characteristics of both powders of *Solanum nigrum* variants (Black and Orange fruited). The shade dried powder of black fruited are dark green in colour and orange fruited are pale green in colour. There is no difference in odour and taste among the powders.

Powder of *Solanum nigrum* (Black fruited) as such is dark green in colour. It did not show any colour difference in both visible and UV light. Powder of *Solanum nigrum* (Orange fruited) as such is pale green in colour. It shows colour difference in both visible and UV light. It was pale green in visible light. In UV light it was green in colour.

Fluorescent behaviour of powder of black fruited *Solanum nigrum* is given in Table 2. When the powders were treated with various chemical reagents like 1 N NaOH in water and ethanol, the powders did not show any colour change in visible and UV light. 1 N HCl shows the brown colour in visible light, it was fluorescent green colour in UV light. The same result was noted in 50% HCl also. The powder with 50% H<sub>2</sub>SO<sub>4</sub> shows greenish brown colour in visible light and fluorescent green in UV light. The results of fluorescent behaviour of powder of *Solanum nigrum* L. variant Orange fruited are presented in Table 3. The powder as such in visible light is pale green in colour where as in UV light it is green colour. Powder with different chemical reagents shows remarkable colour changes in visible as well as UV light in all tested reagents.

**Table 1.** Organoleptic characteristics of whole plant powders of *Solanum nigrum* L. variants.

Sl. No.	Characters	Powders	
		Black fruited	Orange fruited
1.	Colour	Dark green	Pale green
2.	Odour	Characteristic and agreeable odour	Characteristic and agreeable odour
3.	Taste	Bitter	Bitter

**Table 2.** Fluorescent behaviour of Powder of *Solanum nigrum* L. variant black fruited to different chemical reagents.

S. No.	Reagent used	Powder	
		Visible light	UV light
1.	Powder as such	Dark green	Dark green
2.	Powder + 1 N NaOH in water	Fluorescent green	Fluorescent green
3.	Powder + 1 N NaOH in Ethanol	Fluorescent green	Fluorescent green
4.	Powder + 1 N HCl	Brown	Fluorescent green
5.	Powder + 50% HCl	Brown	Fluorescent green
6.	Powder + 50% H <sub>2</sub> SO <sub>4</sub>	Greenish Brown	Fluorescent green

**Table 3.** Fluorescent behaviour of Powder of *Solanum nigrum* L. variant orange fruited to different chemical reagents.

S. No.	Reagent used	Powder	
		Visible light	UV light
1.	Powder as such	Pale green	Green
2.	Powder + 1 N NaOH in water	Dark Green	Yellowish Green
3.	Powder + 1 N NaOH in ethanol	Fluorescent green	Fluorescent Yellowish green
4.	Powder + 1 N HCl	Brown	Fluorescent green
5.	Powder + 50% HCl	Greenish Brown	Fluorescent green
6.	Powder + 50% H <sub>2</sub> SO <sub>4</sub>	Dark Brown	Greenish Dark Brown

Table 4 shows the results of behaviour of powders of *Solanum nigrum* L. variant black fruited with various chemical reagents. The black fruited powders are dark green colour as such without treatment of any chemicals. There was no colour change noted when the powder was treated with conc. HCl and potassium hydroxide in alcohol (5%). The remaining chemical reagents show remarkable colour changes.

Behaviours of powder *Solanum nigrum* L. variant Orange fruited with various chemical reagents are shown in Table 5. The powder as such is pale green in

colour. The same colour was retained when the powder was treated with KOH in water (5%); whereas, the powders are green in colour when treated with ammonia solution, potassium hydroxide in alcohol and sodium nitro prusside solution. The powder with acetic acid shows blackish green, whereas in ammonium oxalate solution and ferric chloride solution show yellowish green in colour and the other chemical reagents shows the noted colours change. Whole plant powders of both variants showed mostly similar characteristic features when treated with various chemical reagents.

**Table 4.** Behaviour of Powder of *Solanum nigrum* L. variant black fruited to various chemical reagents.

S. No.	Powder + Reagent used	Colour of the powder
1.	Powder as such	Dark Green
2.	Powder + Acetic acid	Greenish Black
3.	Powder + Ammonium oxalate solution	Green
4.	Powder + Ammonia solution	Yellowish Green
5.	Powder + Ferric chloride solution (5%)	Pale Green
6.	Powder + Conc. Hydrochloric acid	Dark Green
7.	Powder + Iodine solution	Brownish Green
8.	Powder + Potassium Hydroxide in alcohol (5%)	Dark Green
9.	Powder + Potassium Hydroxide in water (5%)	Fluorescent Yellowish Green
10.	Powder + Conc. Nitric acid (HNO <sub>3</sub> )	Brick Red
11.	Powder + Picric acid	Fluorescent Green
12.	Powder + Sodium Nitroprusside solution	Pale Green
13.	Powder + Conc. Sulphuric acid (H <sub>2</sub> SO <sub>4</sub> )	Brownish Green

**Table 5.** Behaviour of *Solanum nigrum* L. variant orange fruited to various chemical reagents.

S. No.	Powder + Reagent used	Colour of the powder
1.	Powder as such	Pale Green
2.	Powder + Acetic acid	Blackish Green
3.	Powder + Ammonium oxalate solution	Yellowish Green
4.	Powder + Ammonia solution	Dark Green
5.	Powder + Ferric chloride solution (FeCl <sub>3</sub> ) (5%)	Yellowish Green
6.	Powder + Conc. Hydrochloric acid (HCl)	Dark Black Green
7.	Powder + Iodine solution	Brownish Orange
8.	Powder + Potassium Hydroxide (KOH) in alcohol (5%)	Dark Green
9.	Powder + Potassium Hydroxide (KOH) in water (H <sub>2</sub> O) (5%)	Pale Green
10.	Powder + Conc. Nitric acid (HNO <sub>3</sub> )	Brick Red
11.	Powder + Picric acid	Fluorescent Greenish Yellow
12.	Powder + Sodium Nitro prusside solution	Dark Green
13.	Powder + Conc. Sulphuric acid (H <sub>2</sub> SO <sub>4</sub> )	Brownish Green

## Physicochemical studies

Table 6 shows the physico-chemical and extractive values of *Solanum nigrum*. The values of all the parameters showed distinct differences in both the variants. Total ash (16.9221), acid insoluble ash value (3.3165), sulphated ash value (17.4848) and loss on drying (6.0781) are higher in variant orange fruited. Similarly, alcohol and water solubility percentage is higher in variant black fruited when compared to variant orange fruited. Extractive values of most of the extracts of variant black fruited is higher than that of variant orange fruited. In black fruited the percentage of extractability was maximum in water (26.0%) followed by ethanol (10.5%), petroleum ether (1.44%) and chloroform (1.38%) and hexane (0.16%). Minimum extractive value was recorded in hexane (0.16%). In orange fruited the maximum extractive value was recorded in water (18.92%), ethanol (9.55%), petroleum

ether (1.32%), chloroform (1.19%) and hexane (0.24%). While values of loss on drying, total ash value, acid insoluble ash and sulphated ash are higher in orange fruited variant, solubility percentage and extractive value in all the solvents was higher in black fruited variants.

Table 7 shows organoleptic characters of successive solvents extracts of *Solanum nigrum* variants. Dark green colour was observed in petroleum ether solvent in both the variants; whereas, green colour was observed in chloroform and dark brown colour in water. Other solvents such as hexane and ethanol showed remarkable colour changes. In all successive solvents the odour remains same. As far as the taste in concern, chloroform tastes pungent in black fruited variant of *Solanum nigrum*, whereas slightly bitter in orange fruited variant. The rest are more or less identical in taste.

**Table 6.** Physicochemical and extractive values of *Solanum nigrum* L. variants.

S.No.	Characters	Values in %	
		Black fruited	Orange fruited
1.	Loss on drying at 105°C	3.5600	6.0781
2.	Total ash value	14.1146	16.9221
3.	Acid insoluble ash value	1.7124	3.3165
4.	Sulphated ash value	15.2612	17.4848
5.	Solubility percentage		
	a) Alcohol	17.9634	12.6564
	b) Water	23.3861	19.6379
6.	Extractive values in		
	a) Petroleum ether	1.44	1.32
	b) Hexane	0.16	0.24
	c) Chloroform	1.38	1.19
	d) Ethanol	10.5	9.55
	e) Water	26.0	18.92

**Table 7.** Organoleptic characters of successive solvent extracts of *Solanum nigrum* variants.

S. No.	Extracts	Black fruit			Orange fruit		
		Colour	Odour	Taste	Colour	Odour	Taste
1.	Petroleum ether	Dark Green	Characteristic Odour	Bitter	Dark Green	Characteristic Odour	Bitter
2.	Hexane	Dark Green	Characteristic Odour	Slightly Bitter	Pale Green	Characteristic Odour	Bitter
3.	Chloroform	Green	Characteristic Odour	Pungent	Green	Characteristic Odour	Slightly Bitter
4.	Ethanol	Brown	Sharp Odour	Bitter	Dark Brown	Sharp Odour	Strong Bitter
5.	Water	Dark Brown	Sharp Odour	Strong Bitter	Dark Brown	Sharp Odour	Strong Bitter

### Qualitative phytochemical studies

Presence of carbohydrate, alkaloid, fixed oils and fats, saponins, tannins and phenols, gums and mucilage and phytosterols were screened qualitatively in the successive extracts of *Solanum nigrum* variants. The results are summarized in Tables 8 and 9. Presence of carbohydrates was noted only in ethanol extract of variant black fruited and was noted only in water extracts in variant orange fruited. All the extracts except petroleum ether showed presence of alkaloid in both variants. The ethanol and water extracts showed positive result for saponins. The water extracts of both variants showed positive results for gums and mucilage. Tannins and phenols are noted only in ethanol and water extract of variant black fruited but in variant orange fruited,

only the water extract showed the positive results.

### Quantitative phytochemical studies

Total alkaloids, flavonoids, terpenoids and glycosides were estimated. Obtained results were presented in the Table 10. The variant black fruited *Solanum nigrum* showed higher amount of all the above phytoconstituents than that of orange fruited variant. The alkaloids content in both the variants were found to be higher than that of rest of the phytochemicals quantified. All the black fruited extracts showed terpenes positive except in water whereas in all the orange fruited extracts showed positive except in ethanol and water. Both the variants showed positive for fixed oils and fats in petroleum ether and hexane.

**Table 8.** Qualitative Phytochemical screening of *Solanum nigrum* L. variant black fruited.

S. No.	Test for	Test applied / Reagent used	PetE	Hex	CHI	Eth	Wat
1.	Carbohydrate	a) Fehling's	-	-	-	+	-
		b) Benedict's	-	-	-	-	+
2.	Alkaloids	a) Dragendorff's	-	+	+	+	+
		b) Mayer's	-	+	+	+	+
		c) Wagner's	-	+	+	+	+
		d) Hagers	-	+	+	+	+
3.	Fixed oils and fats	Spot test	+	+	-	-	-
4.	Saponins	Foam test	-	-	-	+	+
5.	Tannins and phenols	a) 10% lead Acetate solution	-	-	-	+	+
		b) Ferric chloride solution	-	-	-	-	-
		c) 1% Gelatine containing 10% sodium chloride	-	-	-	-	-
6.	Gums and mucilage	Alcoholic precipitation	-	-	-	-	+
7.	Sterols/ Phytosterol/ Terpenes	Lieberman Burchard	+	+	+	+	-

(+) - present; (-) - completely absent; PetE-Petroleum ether; Hex-Hexane; CHI-Chloroform; Eth-Ethanol; Wat-Water.

**Table 9.** Qualitative Phytochemical screening of *Solanum nigrum* L. variant orange fruited extracts.

S. No.	Test for	Test applied / Reagent used	PetE	Hex	CHI	Eth	Wat
1.	Carbohydrate	a) Fehling's	-	-	-	+	-
		b) Benedict's	-	-	-	+	-
2.	Alkaloids	a) Dragendorff's	-	+	+	+	+
		b) Mayer's	-	+	+	+	+
		c) Wagner's	-	+	+	+	+
		d) Hagers	-	+	+	+	+
3.	Fixed oils and fats	Spot test	+	+	-	-	-
4.	Saponins	Foam test	-	-	-	+	+
5.	Tannins and phenols	a) 10% lead Acetate solution	-	-	-	+	+
		b) Ferric chloride solution	-	-	-	-	-
		c) 1% Gelatine containing 10% sodium chloride	-	-	-	-	-
6.	Gums and mucilage	Alcoholic precipitation	-	-	-	-	+
7.	Sterols/ Phytosterol/ Terpenes	Lieberman Burchard	+	+	+	-	-

(+) - present; (-) - completely absent; PetE-Petroleum ether; Hex-Hexane; CHI-Chloroform; Eth-Ethanol; Wat-Water.

**Table 10.** Quantitative phytochemical analysis of *Solanum nigrum* L. variants.

S. No.	Name of the constituents	Values in mg/g	
		Black fruited	Orange fruited
1	Alkaloid	48.61 ± 0.83	35.65 ± 0.95
2	Flavonoids	23.51 ± 0.88	13.30 ± 0.19
3	Terpenoids	26.76 ± 0.73	11.16 ± 0.73
4	Glycosides	21.02 ± 0.83	10.42 ± 0.19

## Discussion

Phytochemical is a natural bioactive compound found in plants that work with nutrients and fibers to act as a defense system against disease or more accurately, to protect against disease. Some of the most important bioactive phytochemical constituents are alkaloids, essential oils, flavonoids, tannins, terpenoids, saponins, phenolic compounds and many more (Edeoga et al., 2005).

Organoleptic investigation of whole plant powders of *Solanum nigrum* variants exhibited characteristic agreeable odour and bitter taste (Table 1). The fresh aerial parts of the plant are green colour. The successive solvent extracts of *Solanum nigrum* showed dark green to dark brown with characteristic odour. Organoleptic profile is one of the many diagnostic parameters in the proper identification of raw materials (Parvathy and Gopalakrishnan, 1991; Shanmugavadivu and Subramanian, 2009).

Fluorescence analysis of the present study revealed that the powder of *Solanum nigrum* black fruited as such showed fluorescent behaviour (Tables 2 and 3). On treating with various reagents, it has shown characteristic variations (Tables 4 and 5). Analytical tests based on fluorescence in day light and UV light can be used to check identity of powdered drugs (Kokoshi et al., 1958). Fluorescence profiles denote different phytochemical components that are useful in the assessment of active constituents of a drug, responsible for their pharmacological action that is useful for the preparation of genuine Ayurvedic drugs. Similar fluorescence profiling had been undertaken by Chase and Pratt (1949), Kokoshi et al. (1958), Chitra and Thoppil (2002).

Reports of the present investigation revealed that the plant samples contain 3.5600% moisture content in variant black fruited and 6.078% in variant orange fruited (Table 6) (loss on drying). It determines the water drying off from the drug and is used for substance appearing to contain water as the major constituent. The ash content found in the present study sample was presented in Table

6 on dry weight basis. To determine ash content, the plant material is burnt and the residual ash is measured as total and acid-insoluble ash.

In the present study, extractive values of plant powders of *Solanum nigrum* variants revealed that the percentage of extractability was maximum in water (26.0%) followed by ethanol (10.5%), petroleum ether (1.44%) and chloroform (1.38%) and hexane (0.16%). Minimum extractive value was recorded in hexane (0.16%). The maximum extractive value was recorded in water (18.92%), ethanol (9.55%), petroleum ether (1.32%), chloroform (1.19%) and hexane (0.24%) in the black and orange fruited variants respectively. Different extractive value and chemical group tests of different solvent extracts show a path for isolation of different active constituents present in the extracts. Extractive value profiles help in the detection of adulterants during the process of authentication of crude and raw drug materials. Earlier works had revealed extractive value profiles in several medicinal plants (Sharma and Habib, 1995; Khaton et al., 2006).

Qualitative screening in the present study revealed the presence of carbohydrate, alkaloid, fixed oils and fats, saponins, tannins and phenols, gums and mucilage and phytosterols. Preliminary qualitative test according to Mallikharjunah et al. (2007) is useful in the detection of bioactive principles and subsequently may lead to drug discovery and development (Tables 8 and 9). Similar results were also reported by Djaafar and Ridha (2014) in *Solanum nigrum* the Algerian Desert. The data obtained in the present work confirm the relatedness of the investigated *Solanum nigrum* species, as well as reveal their potentials in the drug industry. Similar results were also reported by Elias et al. (2013) who undertook pharmacognostical, phytochemical and anthelmintics activity on leaves of *Solanum nigrum* and discovered the presence of tannins, alkaloids and flavonoids in all the samples.

The present study revealed the presence of the subject secondary metabolites in the variants (Table 10) which is taxonomically useful (Parekh et al., 2005) and it also

brings to bare the fact that the study plant is potential sources of these important phytochemicals.

### Conflict of interest statement

Authors declare that they have no conflict of interest.

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